CORRECTED VERSION

(19) World Intellectual Property Organization International Bureau





(43) International Publication Date 28 October 2004 (28.10.2004)

PCT

(10) International Publication Number WO 2004/092244 A3

(51) International Patent Classification7:

C08G 59/68

(21) International Application Number:

PCT/EP2004/050503

(22) International Filing Date:

13 April 2004 (13.04.2004)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

0697/03

16 April 2003 (16.04.2003) CH

(71) Applicant (for all designated States except US): HUNTS-MAN ADVANCED MATERIALS (SWITZERLAND) GMBH [CH/CH]; Klybeckstrasse 200, CH-4057 Basel (CH).

- (72) Inventor; and
- (75) Inventor/Applicant (for US only): WEIDMANN, Ulrich [DE/CH]; Habsburgerstrasse 38, CH-4055 Basel (CH).
- (74) Agent: MAUÉ, Paul Georg; Solvias AG, Patents, WKL-402.3.04, Klybeckstrasse 191, CH-4002 Basel (CH).
- (81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE,

KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.

(84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

with international search report

(88) Date of publication of the international search report:
23 December 2004

(48) Date of publication of this corrected version:

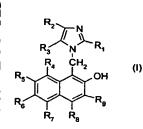
10 November 2005

(15) Information about Correction:

see PCT Gazette No. 45/2005 of 10 November 2005, Section II

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: ACCELERATOR SYSTEMS FOR LOW-TEMPERATURE CURING OF EPOXY RESIN COMPOSITIONS



(57) Abstract: Composition comprising as component A) a 1-imidazoly[mothyl-substituted 2-naphthol compound of the general formula (I) where R₁, R₂ and R₃ each independently of one another are H; C₁₋₁₇alkyl; C₃₋₁₂cycloalkyl, optionally substituted by C₁₋₄alkyl groups; C₄₋₂₀cycloalkyl-alkyl, optionally substituted by C₁₋₄alkyl groups; C₆₋₁₀aryl, optionally substituted by 1-3 C₁₋₄alkyl groups; C₃₋₁₇alkenyl; C₃₋₁₂alkynyl; or aromatic or aliphatic C₃₋₁₂acyf; R₄, R₅, R₆, R₇, R₈, and R₉ each independently of one another are H; C₁₋₁₂alkyl; C₃₋₁₂cycloalkyl, optionally substituted by C₁₋₄alkyl groups; C₄₋₂₀cycloalkyl-alkyl, optionally substituted by C₁₋₄alkyl groups; C₄₋₂₀cycloalkyl-alkyl, optionally substituted by C₁₋₄alkyl groups; C₆₋₁₀aryl, optionally substituted by 1-3 C₁₋₄alkyl groups; C₇₋₁₅phenylalkyl, optionally substituted by 1-3 C₁₋₄alkyl

groups; C₃₋₁₇alkenyl; C₃₋₁₂alkynyl; C₁₋₁₂alkoxy; or OH; and as component B) a phenol which is liquid at room temperature, the weight ratio of component A) to component B) being from 10:90 to 80:20. as accelerator for curable epoxy resin compositions which are used as a compression moulding compound, sinter powder, encapsulating system, casting resin, or for producing prepregs and laminates having very good interlaminar shear strength values using impregnating methods or injection methods, for producing components, especially components of large surface area.

WO 2004/092244 PCT/EP2004/050503

Accelerator systems for low-temperature curing

The invention relates to novel Mannich bases based on modified 1-Imidazolylmethyl-substituted 2-naphthol compounds and also to their use as accelerators for epoxy resin systems which allow high ILS values in laminates, particularly for impregnation by the wet layup method and other impregnating methods. The accelerators described are additionally suitable for sinter powder, casting resin and compression moulding compound applications conducted at temperatures below 200°C, in particular in the temperature range 150°C to 180°C.

The compound 1-imidazolylmethyl-2-naphthol and other imidazole catalysts and accelerators in connection with epoxy resins are known.

The customarily used imidazoles such as imidazole, 1-methylimidazole, 2-methylimidazole, 2-ethyl-4-methylimidazole or else 2-phenylimidazoles, however, in many epoxy resin formulations produce inadequate storage stabilities of prepregs (standing times for short) at room temperature when they are employed in prepreg formulations.

In the past attempts have been made to solve this problem by looking to reduce the reactivity of the imidazoles by formation of salts with organic or inorganic acids: see US 3,356,645 and US 5,001,212, for example. Although improvements in the standing times were achieved in those cases, they are still not sufficient for many applications.

Another way of increasing the standing times is to form Imidazole complexes by reacting imidazoles with metal salts: see US 4,101,514 and US 4,487,914, for example. Generally speaking, the improvement in the standing times that can be achieved in this way is obtained at the expense of an increase in the processing temperatures. Moreover, the metal complexes present in the cured epoxy resin system lead to a deterioration in the dielectric values and also to an increase in the water absorption. In many applications, however, it is required that there is no substantial change in water absorption, since otherwise the glass transition temperature may be lowered, which can lead to a considerable change in the mechanical, electrical and thermal properties of the impregnated component.

EP 0 761 709 describes 1-imidazolylmethyl-substituted 2-naphthol compounds as catalysts which make it possible substantially to avoid the disadvantages described in the above citations. The compounds in question are stable Mannich bases which in epoxy resin

systems lead to a markedly improved standing time of the overall system at room temperature. A formulation comprising such a catalyst can be cured rapidly in the temperature range between 110°C and 150°C. Materials of this kind exhibit good mechanical properties with relatively high glass transition ranges. Prepregs comprising such catalysts can be stored without problems for up to 16 days at room temperature and processed to laminates.

Established methods for producing favourably priced components of large surface area include the wet layup method and other impregnating methods. For reasons of cost, the aim is for temperatures below 100°C during the impregnating operation. For reasons of greater ease of handling the semi-finished products manufactured in this way (prepregs) ought to have a relatively long storage stability at room temperature, which means that the prepreg must be capable of troublefree conversion to the laminate after storage for four days.

Where laminates produced in this way are employed in energy-producing installations, the laminates being subject to rotational movements and shear forces during their use, a certain minimum of adhesion is necessary between the individual layers which make up the laminate. One measure of this adhesion is that known as interlaminar shear strength, also called ILS for short, which is determined in accordance with the ASTM standard (ASTM D 2344-84). A maximum ILS value is therefore an aim for such applications.

It has now been found that 1-(imidazolyl-2-methyl)-2-naphthol does enable outstanding storage stabilities at room temperature in prepreg formulations but not very high interlaminar shear strength values. The maximum achievable value is approximately 22 MPa, irrespective of whether curing is carried out at 60°C for four hours, at 75°C for four hours, at 75°C for eight hours or at 140°C for 30 minutes.

It has now surprisingly also been found that certain compositions of 1-imidazolyl-2-methyl-substituted 2-naphthol compounds with phenols have a profile of properties which allows their advantageous use as accelerators for epoxy resin systems particularly in the context of the wet layup method and other impregnating methods. In particular it is possible in this way to achieve increases in the ILS value of up to 50 MPa.

The Invention accordingly first provides compositions comprising as component A) a 1-imidazolylmethyl-substituted 2-naphthol compound of the general formula (I)

$$\begin{array}{c} R_2 \\ R_3 \\ N \\ R_4 \\ CH_2 \\ CH_2 \\ OH \\ R_8 \\ R_7 \\ R_8 \end{array} \qquad (I),$$

where

 $R_1,\,R_2$ and R_3 each independently of one another are H; $C_{1\text{--}17}alkyl;$

C₃₋₁₂cycloalkyl, optionally substituted by C₁₋₄alkyl groups;

C₄₋₂₀cycloalkyl-alkyl, optionally substituted by C₁₋₄alkyl groups;

 $C_{\text{8-10}}$ aryl, optionally substituted by 1-3 $C_{\text{1-4}}$ alkyl groups;

 C_{7-15} phenylalkyl, optionally substituted by 1-3 C_{1-4} alkyl groups;

 C_{3-17} alkenyl; C_{3-12} alkynyl; or aromatic or aliphatic C_{3-12} acyl;

 $R_4,\,R_5,\,R_8,\,R_7,\,R_8,$ and R_9 each independently of one another are H; $C_{1\text{-}12}alkyl;$

C₃₋₁₂cycloalkyl, optionally substituted by C₁₋₄alkyl groups;

 $C_{4\text{-}20}$ cycloalkyl-alkyl, optionally substituted by $C_{1\text{-}4}$ alkyl groups;

C₆₋₁₀aryl, optionally substituted by 1-3 C₁₋₄alkyl groups;

C₇₋₁₅phenylalkyl, optionally substituted by 1-3 C₁₋₄alkyl groups;

C₃₋₁₇alkenyl; C₃₋₁₂alkynyl; C₁₋₁₂alkoxy; or OH; and

as component B) a phenol which is liquid at room temperature (RT = 15 to 35°C), the weight ratio (in % by weight) of component A) to component B) being from 10:90 to 80:20, preferably from 20:80 to 70:30, more preferably from 25:75 to 50:50.

As component A) preference is given to compounds of the general formula (I) for which the radicals R_1 , R_2 and R_3 are each independently of one another H;

 C_{1-12} alkyl; phenyl; or C_{7-15} phenylalkyl, optionally substituted by 1-3 C_{1-4} alkyl groups; particular preference is given to compounds where R_2 and R_3 are each H; and R_1 is C_{1-12} alkyl; phenyl; or C_{7-15} phenylalkyl, optionally substituted by 1-3 C_{1-4} alkyl groups.

Particular preference as component A) is given to compounds of the general formula (I) for which the radicals R_2 to R_9 are a hydrogen atom and the radical R_1 is C_{1-4} alkyl (methyl, ethyl, n,i-propyl, n,i,t-butyl), or phenyl, optionally substituted by 1-3 C_{1-4} alkyl groups.

As component B) it is preferred to use 1,4-n-pentyl-, -n-hexyl-, -n-heptyl-, -n-octyl-, -n-nonyl-, and -n-decylphenol, and particular preference to using an O,O'-diallyl-bisphenol A.

The fraction of naphthol derivative ought preferably to be at least 20% by weight in the composition of components A) and B) in order to avoid substantial reductions in the glass transition range of the cured formulation as a result of the phenol.

To prepare the compositions of the invention first of all a 1-imidazolylmethyl-substituted 2-naphthol is prepared as described, for example, in EP 0 761 709. Then advantageously the desired phenol (component B) is introduced and the naphthol is added to it, and the mixture is stirred intimately at an elevated temperature of 160°C, for example, for from four to 6 hours. Depending on the choice of the components and of the chosen mixing ratio the products are viscous to high-viscosity liquids or solids. The nature of the composition obtained may also be described as a solid solution of the naphthol in the phenol. Where at least about 20% by weight of a phenol is added to the naphthols, the change in the profile of properties of the naphthol thus "modified" is already so significant that compositions of this kind have a profile of properties which is sufficient for advantageous use in prepregs. Compositions which have been found to be particularly advantageous are those having a slight to marked excess (in % by weight) of phenol.

As mentioned at the outset, the compositions of the invention are suitable as accelerators for curable epoxy resin systems.

The invention therefore further provides curable epoxy resin compositions comprising

- a) an epoxy resin whose epoxide content is from 0.1 to 11, preferably from 0.1 to 2.2, epoxide equivalents/kg,
- b) a composition comprising as component A) a 1-imidazolylmethyl-substituted 2-naphthol compound of the above general formula (I) and as component B) a phenol which is liquid at room temperature (RT = 15 to 35°C) such as, for example, n-pentyl-, n-hexyl-, n-heptyl-, n-octyl-, n-nonyl-, n-decylphenol, in particular an O,O'-diallyl-bisphenol A, the weight ratio (% by weight) of component A) to component B) being from 10:90 to 80:20, preferably from 20:80 to 70:30, more preferably from 25:75 to 50:50,
- c) a curing agent for the epoxy resin, and optionally
- d) an additive customary in epoxy resin technology.

In principle all epoxy resins are suitable as component (a).

Suitable examples include diglycidyl or polyglycidyl ethers of cycloaliphatic polyols, such as 2,2-bis(4'-hydroxycyclohexyl)propane, diglycidyl or polyglycidyl ethers of polyhydric phenols, such as resorcinol, bis(4'-hydroxyphenyl)methane (bisphenol F), 2,2-bis(4'-hydroxyphenyl)propane (bisphenol A), 2,2-bis(4'-hydroxy-3',5'-dibromophenyl)propane, 1,1,2,2tetrakis(4'-hydroxyphenyl)ethane, or condensation products of phenols with formaldehyde, such as phenol novolaks and cresol novolaks; additionally, di- or poly(ß-methylglycidyl) ethers of the above-cited polyalcohols and polyphenols; polyglycidyl esters and poly(ß-methylglycidyl) esters of polybasic carboxylic acids such as phthalic acid, terephthalic acid, tetrahydrophthalic and hexahydrophthalic acid; glycidyl derivatives of aminophenols, such as triglycidyl-p-aminophenol; N-glycidyl derivatives of amines, amides and heterocyclic nitrogen bases, such as N,N-diglycidylaniline, N,N-diglycidyltoluidine, N,N,N',N'-tetraglycidylbis(4-aminophenyl)methane, triglycidyl isocyanurate, N,N-diglycidyl-N,N'-ethyleneurea, N,N'-diglycidyl-5,5-dimethylhydantoin, N,N'-diglycidyl-5-isopropylhydantoin, N,N'-diglycidyl-5,5-dimethyl-6-isopropyl-5,6-dihydrouracil; polyfunctional epoxy resins, such as the 2,6-disubstituted 4-epoxypropylphenyl glycidyl ethers and adducts thereof that are described in EP-A 205 409 and EP-A 204659; bisphenols such substituted with in each case two glycidyloxy groups and 2,3-epoxypropyl groups, such as the 2,2-bis(3'-epoxypropyl-4'-epoxypropylphenyl)propane described in GB 828364; glycidyl derivatives of tetramethylolsubstituted cyclohexanols, cyclohexanones, cyclopentanols and cyclopentanones, such as the compounds described in US 4,549,008; glycidyloxy-substituted benzophenones; and glycidyloxydiketones, such as the compounds described in US 4,649,181.

In general it is also possible to use mixtures of two or more epoxy resins as components in the formulations of the invention.

Suitable epoxy resins include preferably glycidyl ethers such as bisphenol A or F, glycidyl esters, N-glycidyl and N,O-glycidyl derivatives of aromatic or heterocyclic compounds, and also cycloaliphatic glycidyl compounds. They preferably have a functionality of from 0.1 to 2.2 epoxide equivalents/kg.

As curing agents, or component (c), it is possible in principle to use all of the curing agents which are customary in epoxy resin chemistry, such as amines, dicyandiamide, cyanoguanidines, melamines, novolaks, including cresol-novolaks, polyols and anhydrides, for example.

As curing agent it is preferred to use amines and polyamines, examples being those of the Jeffamine type, and others. Examples that may be mentioned include o-, m-, and p-phenylenediamine; diaminotoluenes, such as 2,4-diaminotoluene, 1,4-diamino-2-methoxybenzene, 2,5-diaminoxylene, 1,3-diamino-4-chlorobenzene, 4,4'-diaminodiphenylmethane, 4,4'-diaminodiphenyl ether, 4,4'-diaminodiphenyl thioether, 4,4'-diaminodiphenyl sulfone, 2,2'-diaminobenzophenone, 1,8- or 1,5-diaminonaphthalene, 2,6-diaminopyridine, 1,4-piperazine, 2,4-diaminopyrimidine, 2,4-diamino-s-triazine, di-, tri-, tetra, hexa-, hepta-, octa-, and decamethylenediamine, 3-methylheptamethylene-1,6-diamine, 3-methoxyhexa-methylenediamine, 2,11-diaminododecane, 2,2,4- and 2,4,4-trimethylhexamethylene-diamine, 1,2-bis(3-aminopropoxy)ethane, N,N'-dimethylethylenediamine, N,N'-dimethyl-1,6-diaminohexane and also the diamines of the formulae

 $H_2N(CH_2)_3O(CH_2)_2O(CH_2)_3-NH_2$ and $H_2N(CH_2)_3S(CH_2)_3NH_2$, 1,4-diaminocyclohexane, 1,4-bis(2-methyl-4-aminopentyl)benzene, 1,4-bis(aminomethyl)benzene.

Additionally suitable amines are carbocyclic-aromatic diamines, especially substituted dinuclear diamines, such as bis(3,5-diisopropyl-4-aminophenyl)methane, bis(2-chloro-3,5-diethyl-4-aminophenyl)methane, bis(3-ethyl-4-amino-5-sec-butylphenyl)methane, bis(2-chloro-3,5-diethyl-4-aminophenyl)methane and bis(3,5-diethyl-4-aminophenyl)methane, for example.

Additionally suitable are propane-1,3-diamine, m-xylenediamine, bis(4-aminocyclohexyl)-propane, 3-aminomethyl-3,5,5-trimethylcyclohexylamine (isophoronediamine), polyaminoamides, examples being those consisting of aliphatic polyamines and dimerized or trimerized fatty acids; polyphenols, such as resorcinol, hydroquinone, bisphenol A and phenol/aldehyde resins, and also polythiols such as "Thiokols" for example.

Particular preference is given to diamines or polyamines, amino-terminated polyalkylene glycols and polyoxypropylenediamines (e.g. Jeffamines, in this case Jeffamine D 230, amino-poly-THF) or polyaminoamides, especially dimers or copolymers of propylene glycol and ethylene glycol, amino-terminated polybutadienes with molecular weights in the range from about 150 to 5000, in particular from 200 to 600.

As optional component d) it is possible to use additives customary in epoxy resin technology. By these are meant the customary auxiliaries and additives which are known to and used by the person skilled in the art in respect of the particular application. Examples include organic and inorganic fillers and pigments, release agents, and viscosity-influencing additives.

To prepare the curable compositions of the invention it is possible for the accelerators of the invention to be dissolved beforehand in the curing agent, at generally elevated temperatures: for example, when using a Jeffamine, at about 80°C. Solutions of this kind can be cooled to 40°C and then mixed with the epoxy resin. These mixtures can then be used directly as impregnating solutions. Another possibility is to disperse the compositions of the invention homogeneously in the epoxy resin beforehand, by means for example of suitable stirrers, such as an Ultra-Turrax or a triple roll mill.

The compounds of the invention are used advantageously at from 5 to 40 parts by weight, preferably from 5 to 30 parts by weight, based on the overall formulation comprising epoxy resin, curing agent, accelerator and, where used, additives. It is particularly advantageous to use from 5 to 20 parts by weight. The curing agents are employed in the customary amounts, which are therefore to be calculated such that on average per epoxide group there are from 0.5 to 1.5, preferably from 0.8 to 1.2, functional groups of the curing agent. Auxiliaries and additives can be used in principle in wide quantity ranges, provided that this is possible without a significant increase in the viscosity of the desired compositions.

Prepregs comprising inventive accelerator systems of this kind have the capacity to give an ILS value of up 50 MPa. Accordingly the accelerators of the invention are particularly suitable for use in epoxy resin formulations which are employed as compression moulding compounds, sinter powders, encapsulating systems, casting resins and for producing prepregs and laminates by the wet layup method and injection methods, especially for producing components of large surface area.

Experimental section

a) Preparation of a composition comprising naphthol component and phenol component

The phenol component is charged to a vessel and then the naphthol component is added. The amounts used can be seen from Table 1 below. The mixture is stirred at 160°C for 4 hours. This gives viscous to high-viscosity liquids of dark red to black coloration or solids.

Table 1:

Experiment	1	2	3	4
Naphthol 1) [% by weight]	10	30	50	70
Phenol 2) [% by weight]	90	70	50	30
State	liquid	liquid	solid	solid
Viscosity η _{60°C} [Pa•s]	ng ³⁾	28	6000	nm ₃₎

- 1-Imidazolymethyl-2-naphthol from Vantloo AG);
- Diallylbisphenol A (from Vantico AG);
- ng = not measured, nm = not measurable at 60°C, viscosity determined with Rheometrix RD2 with a plate/plate setup

b) Use examples of the accelerator formulations of the invention

The products from the above experimental section a) are dissolved at temperatures between 60°C and 80°C in the amine curing agent (in this case Jeffamine® D-230). This gives dark brown clear solutions. After cooling to room temperature, this solution is admixed with the calculated amount of epoxy resin. For details see Table 2 below:

Table 2

Experiment	Comparison	5
	Composition:	
LY 556 1)	100	100
Amine 2)	10	10
Accelerator 3)	10	
E2 4)		11.7
Prepre	gs and lamina	tes:
Preparation 5)	8	4
Prepreg ⁶⁾	12	12
Resin content 7)	40-42	40-42
Storage 8)	RT	RT
ILS value a)		
F _{max} [N] ⁹⁾	601+/-38	562+/-31
Σ _{max} [MPa] ¹⁰⁾	22.6+/-1.4	14.1+/-0.7
ILS value b)		
F _{max} [N]	426+/-37	10501+/-40
Σ _{mex} [MPa]	16.0+/-1.4	52.4+/-1.5
ILS value c)		
F _{max} [N]	400+/-33	1096+/-51
Σ _{max} [MPa]	16.0+/-1,3	42.0+/-2.0
ILS value d)		
F _{max} [N]	nm	1256+/-22
σ _{max} [MPa]	nm	46.2+/-1.1

1) Araldite LY 556 Bisphenol A resin (Vantico AG) 2) Amine Jeffamine® D-230 3) Accelerator 1-Imidazolylmethyl-2-naphthol (Vantico AG) ⁴⁾ E2 Experiment 2 (see table 1) 5) Preparation Preparation of the prepregs and laminates after number of days 6) Prepreg Number of plies = 12 7) Resin content Resin content in per cent after the laminates have been pressed 8) Storage At 20°C to 25°C = RT 9) F_{max} Ultimate strength (breaking load) to ASTM D 2344 ¹⁰⁾ σ_{max} Shear strength to ASTM D 2344 ILS value a) Interlaminar shear strength (ILS) after 30 min at 140°C ILS value b) Interlaminar shear strength (ILS) after 4 h at 75°C ILS value c) Interlaminar shear strength (ILS) after 8 h at 75°C ILS value d) Interlaminar shear strength (ILS) after 4 h at 60°C

Surprising are the very good interlaminar shear strengths which can be obtained with the accelerators of the invention in the lower temperature range (60°C-75°C/4-8 hours). The values found are substantially higher than in the case of the comparison system.

Claims

 Composition comprising as component A) a 1-imidazolylmethyl-substituted 2-naphthol compound of the general formula (I)

$$\begin{array}{c} R_2 \\ R_3 \\ N \\ R_4 \\ CH_2 \\ CH_2 \\ OH \\ R_8 \\ R_7 \\ R_8 \end{array} \tag{1}$$

where

 R_1 , R_2 and R_3 each independently of one another are H; C_{1-17} alkyl;

 C_{3-12} cycloalkyl, optionally substituted by C_{1-4} alkyl groups;

 $C_{4\text{--}20}$ cycloalkyl-alkyl, optionally substituted by $C_{1\text{--}4}$ alkyl groups;

C₆₋₁₀aryl, optionally substituted by 1-3 C₁₋₄alkyl groups;

C₇₋₁₅phenylalkyl, optionally substituted by 1-3 C₁₋₄alkyl groups;

C₃₋₁₇alkenyl; C₃₋₁₂alkynyl; or aromatic or aliphatic C₃₋₁₂acyl;

 $R_4,\,R_5,\,R_6,\,R_7,\,R_8,$ and R_9 each independently of one another are H; $C_{1\text{--}12}alkyl;$

 C_{3-12} cycloalkyl, optionally substituted by C_{1-4} alkyl groups;

C₄-₂ocycloalkyl-alkyl, optionally substituted by C₁-₄alkyl groups;

C₆₋₁₀aryl, optionally substituted by 1-3 C₁₋₄alkyl groups;

C₇₋₁₅phenylalkyl, optionally substituted by 1-3 C₁₋₄alkyl groups;

C₃₋₁₇alkenyl; C₃₋₁₂alkynyl; C₁₋₁₂alkoxy; or OH; and

as component B) a phenol which is liquid at room temperature, the weight ratio of component A) to component B) being from 10:90 to 80:20.

- Composition according to Claim 1, comprising as component A) a compound in which
 the radicals R₁, R₂ and R₃ each independently of one another are H; C₁₋₁₂alkyl; phenyl;
 or C₇₋₁₅phenylalkyl, optionally substituted by 1-3 C₁₋₄alkyl groups.
 - Composition according to Claim 2, comprising as component A) a compound where R₂ and R₃ are each H; and R₁ is C₁₋₁₂alkyl; phenyl; or C₇₋₁₅phenylalkyl, optionally substituted by 1-3 C₁₋₄alkyl groups.

- 4. Composition according to Claim 3, comprising as component A) a compound of the general formula (I) in which the radicals R₂ to R₉ are a hydrogen atom and the radical R₁ is C₁₋₄alkyl, or phenyl, optionally substituted by 1-3 C₁₋₄alkyl groups.
- Composition according to Claim 1, characterized in that as component B), 1,4-n-pentyl-,
 -n-hexyl-, -n-heptyl-, -n-octyl-, -n-nonyl-, -n-decylphenol or O,O'-diallyl-bisphenol A is
 used.
- Composition according to Claim 1, characterized in that the weight ratio of component A) to component B) is from 20:80 to 70:30, preferably from 25:75 to 50:50.
- 7. Use of a composition according to Claim 1 as accelerator in epoxy resin compositions.
- 8. Curable composition comprising
 - a) an epoxy resin whose epoxide content is from 0.1 to 11 epoxide equivalents/kg,
 - b) from 5 to 40 parts by weight, based on the overall composition of components a) to
 d), of a composition according to Claim 1,
 - c) a curing agent for the epoxide resin, calculated such that per epoxide group there are from 0.5 to 1.5 functional groups of the curing agent, and optionally
 - d) an additive customary in epoxy resin technology.
- Composition according to Claim 8, characterized in that the curing agent is selected from amines, preferably from diamines and polyamines.
- Composition according to Claim 7, characterized in that the curing agent is a polyoxypropylenediamine.
- 11. Composition according to Claim 8, characterized in that the epoxy resin is a glycidyl ether, glycidyl ester, N-glycidyl or N,O-glycidyl derivative of an aromatic or heterocyclic compound, or a cycloaliphatic glycidyl compound.

WO 2004/092244 PCT/EP2004/050503

- 12 -

12. Use of a curable composition according to Claim 6 as a compression moulding compound, sinter powder, encapsulating system, casting resin, for producing prepregs and laminates using impregnating methods or injection methods, for producing components, especially components of large surface area.

INTERNATIONAL SEARCH REPORT

Intraconal Application No PC1/EP2004/050503

4 01 4501	EICATION OF CUD IDOT		
IPC 7	FICATION OF SUBJECT MATTER C08G59/68		
According to	o International Patent Classification (IPC) or to both national classific	ration and IPC	
	SEARCHED		
IPC 7	ocumentation searched (classification system followed by classification code)	ion symbols)	
Documenta	tion searched other than minimum documentation to the extent that	such documents are included in the fields s	earched
Electronic d	ata base consulted during the international search (name of data be	ase and, where practical, search terms used	0)
EPO-In	ternal, PAJ, CHEM ABS Data, WPI Data	a	
			•
C. DOCUM	ENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the re-	levant passages	Relevant to claim No.
Α	EP 0 761 700 A (CIDA OFTOV 40)		
^	EP 0 761 709 A (CIBA GEIGY AG) 12 March 1997 (1997-03-12)		1-12
	cited in the application		
	claim 1; examples 1-7		
Α	PATENT ABSTRACTS OF JAPAN		1-12
	vol. 0122, no. 83 (C-518), 3 August 1988 (1988-08-03)		_
	& JP 63 061016 A (NEW JAPAN CHEM	CO LTD).	
	17 March 1988 (1988-03-17)		
	abstract 	·	
Α	EP 0 086 102 A (CELANESE CORP)		1-12
	17 August 1983 (1983-08-17) claims 1,2		
	•	-/	
		,	
X Furti	ner documents are listed in the continuation of box C.	X Patent family members are listed in	n annex.
	tegories of cited documents:	"T" later document published after the inte	rnational filing date
consid	ent defining the general state of the art which is not lered to be of particular relevance	or priority date and not in conflict with cited to understand the principle or the invention	the application but
nung a		"X" document of particular relevance; the cannot be considered novel or cannot	claimed invention
WHICH	ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another	involve an inventive step when the do "Y" document of particular relevance; the o	cument is taken alone
"O" docume	n or other special reason (as specified) ent referring to an oral disclosure, use, exhibition or	document is combined with one or mo	ventive step when the
P docume	neans ent published prior to the international filling date but aan the priority date claimed	ments, such combination being obvious in the art.	us to a person skilled
	actual completion of the international search	*&* document member of the same patent	
_		Date of mailing of the international sea	ion tahou
2	6 October 2004	03/11/2004	
Name and n	nalling address of the ISA European Patent Office, P.B. 5818 Patentlaan 2	Authorized officer	
	NL - 2280 HV Hijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo ni.	010.11.	
	Fax: (+31-70) 340-3016	O'Sullivan, T	



Int: ional Application No	
PCT/EP2004/050503	

C.(Continu	etion) DOCUMENTS CONSIDERED TO BE RELEVANT	PC1/EP2004/050503
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 2002/058756 A1 (KONARSKI MARK M) 16 May 2002 (2002-05-16) sentence 25, paragraph 63	1-12
A	YOSHITOMI PHARMACEUTICAL INDUSTRIES ET AL: "1-(1,2-Dihydronaphthalen-3-ylmethyl or 2-indenylmethyl)imidazoles" CAPLUS, 1981, XP002163356 AN- 1981-587256 abstract	1-12
	,	
	•	
n PCT/ISA/21	0 (continuation of second sheet) (January 2004)	

INTERNATIONAL SEARCH REPORT

information on patent family members

Intermional Application No
PC1/EP2004/050503

Patent document dted in search report		Publication date		Patent family member(s)	Publication date
EP 0761709	Α	12-03-1997	US	5591811 A	07-01-1997
			BR	9603729 A	26-05-1998
			CA	2185207 A	
			DE	69608054 D	
			DE	69608054 T	
			ΕP	0761709 A	
			ES	2146378 T	
			JP	9118741 A	06-05-1997
			TW	487721 B	21-05-2002
JP 63061016	Α	17-03-1988	JP	1812687 C	27-12-1993
			JP	5021926 B	26-03-1993
EP 0086102	A	17-08-1983	CA	1206687 A	1 24-06-1986
			DE	3363857 D	
			EP	0086102 A	
			JP	1611109 C	15-07-1991
			JP	2036129 B	15-08-1990
			JР	58147418 A	02-09-1983
		·=·	US	4417010 A	22-11-1983
US 2002058756	A1	16-05-2002	US	6670430 B	l 30-12-2003
			US	2003190479 A	